M.SC. ENTRANCE QUESTION

SUBJECT: PHYSICS

Q1. A unit vector perpendicular to the plane containing $\vec{A} = \hat{i} + \hat{j} - 2\hat{k}$ and $\vec{B} = 2\hat{i} - \hat{j} + \hat{k}$ is

$$(A) \frac{1}{\sqrt{26}} \left(-\hat{\imath} + 3\hat{\jmath} - 4\hat{k} \right)$$
$$(B) \frac{1}{\sqrt{19}} \left(-\hat{\imath} + 3\hat{\jmath} - 3\hat{k} \right)$$
$$(C) \frac{1}{\sqrt{35}} \left(-\hat{\imath} + 5\hat{\jmath} - 3\hat{k} \right)$$
$$(D) \frac{1}{\sqrt{35}} \left(-\hat{\imath} - 5\hat{\jmath} - 3\hat{k} \right)$$

Q2. The real part of an analytic function is $u(x, y) = x^2 - y^2$. The imaginary part of the function is then

(A) $x^{2} - y^{2} - 2xy$ (B) $x^{2} + y^{2}$ (C) $x^{2} + y^{2} + xy$ (D)2xy Ans: D

Q3. The function $e^{\cos x}$ is Taylor expanded about x = 0. The coefficient of x^2 is

 $(A) -\frac{1}{2}$ $(B) -\frac{e}{2}$ $(C) \frac{e}{2}$ (D) ZeroAns: B

Q4. Let M be a 2x2 matrix. Its trace is 6 and its determinant has value 8. Its eigen values are

(*A*) 2 and 4

(*B*) 3 and 3

(*C*) 2 and 6

(D) - 2 and - 3

Ans: A

Q5. The solution y(x) of the differential equation $y\frac{dy}{dx} + 3x = 0$, y(1) = 0 is described by

(A) an ellipse

- (B) a circle
- (C) a parabola
- (*D*) a straight line

Ans: A

Q6.The radial component of acceleration in plane polar coordinate is given by

(A)
$$\frac{d^2 r}{dt^2}$$

(B) $\frac{d^2 r}{dt^2} - r \left(\frac{d\theta}{dt}\right)^2$
(C) $\frac{d^2 r}{dt^2} + r \left(\frac{d\theta}{dt}\right)^2$
(D) $2 \frac{d\theta}{dt} \frac{d\theta}{dt} + \frac{d^2\theta}{dt^2}$

Ans: B

Q7. Let (x, y) denote the coordinates in a rectangular cartesian coordinate system *C*.

Let (x', y') denote the coordinates in another coordinate system C' defined by

$$x' = 2x + 3y$$
$$y' = -3x + 4y$$

The area element in C' is

- (A) dx' dy'
- $(B) \ 12 dx' dy'$

$$(C) \ \frac{1}{17} dx' dy'$$

- (D) x' dx' dy'
- Ans: C

Q8. The function $f(x) = \frac{8x}{x^2 + 9}$ is continuous everywhere except

- (A) x = 0
- (B) $x = \pm 9$
- $(C) x = \pm 9i$
- $(D)\,x=\pm 3i$

Ans: D

Q9. If $\phi(x, y, z)$ is a scalar function which satisfies the Laplace equation, the gradient of ϕ is

- (A) Solenoidal and irrotational
- (B)Solenoidal but not irrotational
- (C) Irrotational but not solenoidal and
- (D)Neither solenoidal nor irrotational

Ans: A

Q10. The eigen values of
$$\begin{pmatrix} 3 & i & 0 \\ -i & 3 & 0 \\ 0 & 0 & 6 \end{pmatrix}$$
 are

- (*A*) 2,4 and 6
- (*B*) 2*i*, 4*i* and 6
- (*C*) 2*i*, 4 and 8
- (*D*) 0,4 and 8
- Ans: A
- Q11. If the motion of a particle is described by $x = 5\cos(8\pi t)$, $y = 5\sin(8\pi t)$ and z = 5t then the

trajectory of the particle is

- (A) Circular
- (B) Elliptical
- (C) Helical
- (D) Spiral
- Ans: C

Q12. Given z = x + iy, the contour integration $\oint_C \frac{dz}{z}$ is equal to (where *C* is any anticlockwise

contour going around the origin

- $(A) \pi i$
- (B) 2πi
- (C) πi
- $(D) \pi/2$

Ans: B

Q13. An integral is defined to be
$$I = \int_{0}^{\infty} \frac{\sin x}{x} dx$$
, then *I* is equal to

 $(A) - \pi/\cos\sqrt{2}$

(B) $\pi/2$ (C) $-\cos \sqrt{2}/\pi$ (D) $2/\pi$ Ans: B Q14. The solution to the nonlinear differential equation $\frac{df}{dx} + \alpha f^2 = 0$ with boundary condition f(0) = 1 and α is a constant is given by (A) $\cos \alpha x$ (B) $\sin \alpha x / \alpha x$ (C) $(\alpha x + 1)^{-1}$ (D) $(\alpha x + 1)^{-2}$ Ans: C Q15. Let $f(x, y) = x^3 - 2y^3$. The curve along which $\nabla^2 f = 0$ is

(A) $x = \sqrt{2} y$ (B) x = 2y(C) $x = \sqrt{6} y$ (D) x = -y/2Ans: B

Q16. A curve is given by $\vec{r}(t) = t\hat{i} + t^2\hat{j} + t^3\hat{k}$. The unit vector of the tangent to the curve at t = 1 is

$$(A) \frac{\hat{i} + \hat{j} + \hat{k}}{\sqrt{3}}$$
$$(B) \frac{\hat{i} + \hat{j} + 2\hat{k}}{\sqrt{6}}$$
$$(C) \frac{\hat{i} + 2\hat{j} + 2\hat{k}}{3}$$
$$(D) \frac{\hat{i} + 2\hat{j} + 3\hat{k}}{\sqrt{14}}$$

Ans: D

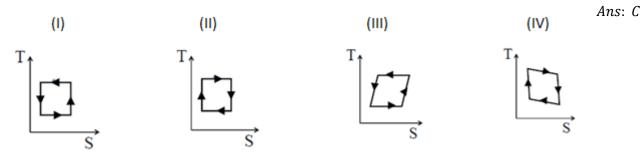
Q17.In a heat engine based on Carnot engine, heat is added to the working subsance at constant

(A) Entropy

(B) Pressure

(C) Temperature

(D) Volume



Q18. Which one of the figures correctly represents the T - S diagram of a Carnot engine?

(A) II

(B) III

(C) I

(D) IV

Ans: B

Q19. Two boxes A and B contain equal number of molecules of the same gas. if the volumes are V_A and V_B

and λ_A and λ_B denote respective mean free paths, then

 $(A) \lambda_A = \lambda_B$ $(B) \ \frac{\lambda_A}{V_A} = \frac{\lambda_B}{V_B}$ $(C) \quad \frac{\lambda_A}{V_A^{1/3}} = \frac{\lambda_B}{V_B^{1/3}}$ $(D) \lambda_A V_A = \lambda_B V_B$

Q20. The equation of state for one mole of a non – ideal gas is given by $PV = A\left(1 + \frac{B}{V}\right)$, where

the coefficient A and B are temperature dependent. if the volume changes from V_1 to V_2 in an isothermal process, the work done by the gas is

 $(A) AB\left(\frac{1}{V_1} - \frac{1}{V_2}\right)$ (B) $AB \ln\left(\frac{V_2}{V_1}\right)$

(C)
$$A \ln \left(\frac{V_2}{V_1}\right) + AB \left(\frac{1}{V_1} - \frac{1}{V_2}\right)$$

(D) $A \ln \left(\frac{V_2 - V_1}{V_1}\right) + B$

Ans: C

- Q21. The rms velocity of oxygen molecule is given by at some temperature T. The molecules of another gas have the same rms velocity at temperature T/16. The second gas is
- (A) Hydrogen
- (B) Helium
- (C) Nitrogen
- (D) Neon

Ans: A

Q22. Isothermal compressibility is given by

 $(A) \frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{T}$ $(B) \frac{1}{P} \left(\frac{\partial P}{\partial V}\right)_{T}$ $(C) - \frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{T}$ $(D) - \frac{1}{P} \left(\frac{\partial P}{\partial V}\right)_{T}$

Ans: C

Q23. A blackbody at a temperature of 6000K emits radiation whose intensity spectrum peaks at 600nm.

If the temperature is reduced to 300K, the spectrum will peak at

(A)12nm

(*B*)120µm

- (C)120 nm
- (D)12µm
- Ans: D
- Q24. An ideal monoatomic gas , initially at $T = 20 \,^{\circ}C$ expands adiabatically from a volume V_0 to $5V_0$.

Then the final temperature is

(*A*) − 20°C

(*B*) − 33 °C

(*C*) − 173°C

 $(D) - 113^{\circ}C$

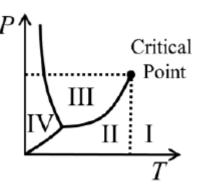
Ans: C

Q25. During free expansion of an ideal gas under adiabatic condition, the internal energy of the gas

(A)Decreases

- (B) Initially decreases and then increases
- (C) Increases
- (D) Remains constant

Ans: D



Q26. In the given phase diagram for a pure substance, region I, II, III, IV respectively represents

- (A) Vapour, gas, solid, liquid
- (B) Gas, vapor, liquid, solid
- (C) Gas, liquid, vapor, solid
- (D) Vapor, gas, liquid, solid

Ans: B

- Q27. A system undergoes a thermodynamic transformation from state S_1 to S_2 via two different paths 1 and 2. The heat absorbed and work done along path 1 are 50*J* and 30*J* respectively. If the heat absorbed along path 2 is 30*J*, the work done along path 2 is
- (A) Zero
- (B) 10J
- (C)20J
- (D)30J
- Ans: B

Q28. The root mean square(rms) speeds of Hydrogen atoms at 500K, V_H , and Helium atoms at 2000K, V_{He} ,

are related as

 $(A) V_{H} > V_{He}$ $(B) V_{H} < V_{He}$ $(C) V_{H} = V_{He}$ $(D) V_{H} \gg V_{He}$

Ans: C

Q29. For the molecules of an ideal gas the ratio of the most probable speed to average speed to root

mean square velocity is given by

$$(A)\sqrt{2}:\sqrt{\pi/8}:\sqrt{3}$$
$$(B)\sqrt{3}:\sqrt{\pi/8}:\sqrt{2}$$
$$(C)\sqrt{3}:\sqrt{8/\pi}:\sqrt{2}$$

$$(D)\,\sqrt{2}{:}\,\sqrt{8/\pi}{:}\,\sqrt{3}$$

Ans: D

Q30. For a mole of ideal gas at T = 35 °C, what is the work done for an isothermal expansion

from a volume V_0 to $10V_0$?

 $(A)6 \ge 10^3 J$

(*B*) $3 \ge 10^3 J$

- $(C) 10^{3} J$
- $(D) 10^4 J$

Ans: A

Q31.Which one of the following is an impossible magnetic field ?

$$(A) \vec{B} = 3x^{2}z^{2}\hat{x} - 2xz^{3}\hat{z}$$

$$(B) \vec{B} = -2xy\hat{x} + yz^{2}\hat{y} + \left(2yz - \frac{z^{3}}{3}\right)\hat{z}$$

$$(C) \vec{B} = (xz + 4y)\hat{x} - yx^{3}\hat{y} + \left(x^{3}z - \frac{z^{2}}{2}\right)\hat{z}$$

$$(D) \vec{B} = -6xz\hat{x} + 3yz^{2}\hat{y}$$

Ans: D

Q32. The expression for the magnetic field that induces the electric field

$$\vec{E} = K(yz\hat{\imath} + 3z\hat{\jmath} + 4y\hat{k})\cos(\omega t) \text{ is}$$

$$(A) - \frac{K}{\omega}(\hat{\imath} + y\hat{\jmath} - z\hat{k})\sin(\omega t)$$

$$(B) - \frac{K}{\omega}(\hat{\imath} + y\hat{\jmath} + z\hat{k})\sin(\omega t)$$

$$(C) - \frac{K}{\omega}(\hat{\imath} - y\hat{\jmath} + z\hat{k})\sin(\omega t)$$

$$(D) - \frac{K}{\omega}(\hat{\imath} + y\hat{\jmath} + z\hat{k})\sin(\omega t)$$

Q33. For an electromagnetic wave travelling in free space, given by $\vec{E} = E_m \sin(\omega t - \beta z)\hat{y}$,

the magnetic field will be given by

$$(A)\vec{B} = -\frac{E_m\beta}{\omega}\sin(\omega t - \beta z)\hat{x}$$
$$(B) - \frac{E_m\beta}{\omega}\cos(\omega t - \beta z)\hat{z}$$
$$(C)\frac{E_m\beta}{\omega}\sin(\omega t - \beta z)\hat{y}$$

(D) None of the above

Ans: A

Q34. The electric field of an electromagnetic field is given by

 $\vec{E} = (2\hat{k} - 3\hat{j})x10^{-3}\sin[10^7(x + 2y + 3z - \beta t)]$. The value of β is (*c* is the speed of light)

(A) $\sqrt{14} c$

 $(B)\sqrt{12} c$

 $(C)\sqrt{10} c$

$$(D)\sqrt{7} c$$

Ans: A

Q35. Three point charges each carrying a charge q are placed on the vertices of an equilateral triangle of side *L*. The electrostatic potential energy of the configuration is

$$(A) \frac{1}{4\pi\epsilon_0} \frac{q^2}{L}$$
$$(B) \frac{2}{4\pi\epsilon_0} \frac{q^2}{L}$$
$$(C) \frac{3}{4\pi\epsilon_0} \frac{q^2}{L}$$
$$(D) \frac{1}{\pi\epsilon_0} \frac{q^2}{L}$$

Ans: C

Q36. For a quantum particle confined inside a cubic box of side L, the grund state energy is given by E_0 .

The energy of the first excited state is

 $(B) \sqrt{2} E_0$

 $(C) 3E_0$

 $(D) \, 6E_0$

Ans: A

Q37. The maximum number of intensity minima that can be observed in the Fraunhofer diffraction pattern of a single slit (width $10 \,\mu m$) illuminated by a laser beam(wavelength 0.630 μm will be

(A) 4

(*B*) 7

(*C*)12

(D) 14

Ans: A

Q38.Light of wavelength λ (in free space) propagates through a dispersive medium with refractive index $n(\lambda) = 1.5 + 0.6\lambda$. The group velocity of a wave travelling inside this medium in units of 10^8 m/s is

 $n(\lambda) = 1.5$ (A) 1.5

(*B*)2.0

(*C*)3.0

(D) 4.0

Ans: B

Q39. Arrange the following telescopes, where D is the telescope diameter and λ is the wavelength,

in order of decreasing resolving power.

I. D = 100m, $\lambda = 21cm$ II. D = 2m, $\lambda = 500nm$ III. D = 1m, $\lambda = 100nm$ IV. D = 2m, $\lambda = 10mm$ (A) III, II, IV, I (B) II, III, I, IV (C) IV, III, II, I, IV (D) III, II, I, IV Ans: B

Q40. A linearly polarised light falls on a quarter wave plate and the emerging light is found to be elliptically polarised. the angle between the fast axis of the quarter wave plate and the plane of polarisation , can be

(A) 30°

(*B*) 45°

(*C*) 90°

 $(D)\,180^\circ$

Ans: A

Q41. The plane of polarisation of a plane polarised light rotates by 60° after passing through a wave plate. The pass axis of the wave plate is at an angle α with respect to the plane of polarisation of the incident light. The wave plate and α are

- (A) $\lambda/4$, 60°
- (B) $\lambda/2$, 30°
- (*C*) $\lambda/2$, 120°
- (*D*) $\lambda/4$, 30°

Ans: C

Q42. An object of density ρ is floating in a liquid with 75% of its volume submerged. The density of the liquid is

$$(A) \frac{4}{3}\rho$$
$$(B) \frac{3}{2}\rho$$
$$(C) \frac{8}{5}\rho$$
$$(D) 2\rho$$

Ans: A

Q43. The moment of inertia of a solid sphere (radius *R* mass *M*)about the axis which is at a distance

of R/2 from the centre is

$$(A) \ \frac{3}{20} MR^{2}$$
$$(B) \ \frac{1}{2} MR^{2}$$
$$(C) \ \frac{13}{20} MR^{2}$$
$$(D) \ \frac{9}{10} MR^{2}$$

Ans: C

Q44. Metallic lithium has bcc crystal structure. each unit cell is a cube of side a. the number of atoms per unit

volume is

$$(A) \frac{1}{a^3}$$
$$(B) \frac{2}{\sqrt{2} a^3}$$
$$(C) \frac{2}{a^3}$$
$$(D) \frac{4}{a^3}$$
$$(D) 0$$
Ans: B

*Q*45. A plane in a cubic lattice makes intercepts of a, a/2 and 2a/3 with the three crystographic axes respectively. the miller indices for this plane are

(A) (2 4 3)

(B)(3 4 2)

(C)(634)

(D) (1 2 3)

Ans: C

Q46.Thermal runway in a transistor biased in the active region is due to

(A) Change in reverse collector saturation current due to rise in temperature.

(B)Breakdown under reverse biasing.

(C) Change in β which increases with temperature.

(D) Base emitter voltage V_{BE} which decreases with rise in temperature.

Ans: A

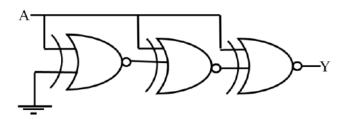
Q47. The Boolean expression $(A + B)(A + \overline{B})(\overline{A} + B)$ is equivalent to

 $(A) A\overline{B}$

 $(B) \overline{A} B$

- $(C) \bar{A}\bar{B}$
- (D) AB

Ans: D



Q48. For the given circuit , the output Y is

(B)	1
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(C) A

 $(D)\,\bar{A}$

Ans: D

Q49. Let T_g and T_e be the kinetic energies of the electron in the ground and the third excited states of a hydrogen atom, respectively. According to the Bohr model, the ratio T_g/T_e

(*A*) 3

(*B*) 4

- (C)9
- (D)16
- Ans: D

Q50. The Hamiltonian of a particle in one dimension is given by $H(x, p) = \frac{p^2}{2m} + \lambda px + \frac{\lambda}{2}x^2$ where

m and λ are constant. the corresponding Lagrangian is

$$(A) L = \frac{m}{2} (\dot{x})^2 - \lambda m x \dot{x} - \frac{\lambda}{2} x^2$$
$$(B) L = \frac{m}{2} (\dot{x} - \lambda x)^2 - \lambda m x \dot{x} - \frac{\lambda}{2} x^2$$
$$(C) L = \frac{m}{2} (\dot{x} - \lambda x)^2 - \frac{\lambda}{2} x^2$$
$$(D) L = \frac{m}{2} (\dot{x})^2 - \frac{\lambda}{2} x^2$$

Ans: C